



Class – X (Going to XI) – Sample Paper
Duration : 2 hrs. | Maximum Marks : 180

IMPORTANT INSTRUCTIONS

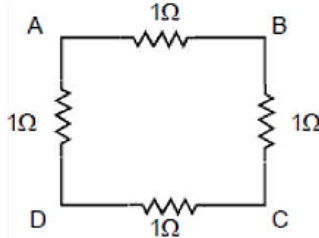
1. This Booklet is your Question Paper. DO NOT break seal of Booklet until the invigilator instructs to do so.
2. Fill your APRE Roll No. & Answer Sheet No. in the space provided on the cover page.
3. Please make sure that paper you received is of your class only.
4. The Answer Sheet is provided to you separately which is a machine readable Optical Response Sheet (ORS).
You have to mark your answers in the ORS by darkening bubble, as per your answer choice, by using black or blue ball point pen.
5. After breaking the Question Paper seal, check there are 8 pages in the booklet. This Question Paper contains 60 MCQs with 4 choices (Subjects: Physics: 20, Chemistry: 20, Biology: 20)
6. Think wisely before darkening bubble as there is negative marking for wrong answer. Answer once marked by pen cannot be cancelled.
7. Marking Scheme:
 - a. If darkened bubble is RIGHT answer: 3 Marks.
 - b. If darkened bubble is WRONG answer: 1 Mark (Minus One Mark).
 - c. If no bubble is darkened in any question: No Mark.
8. If you are found involved in cheating or disturbing others, then your ORS will be cancelled.
9. Do not put any stain on ORS and hand. It over back properly to the invigilator.

Name of the Candidate: _____

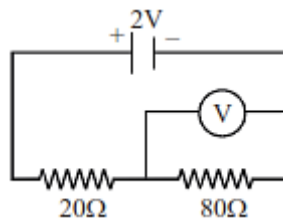
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PHYSICS

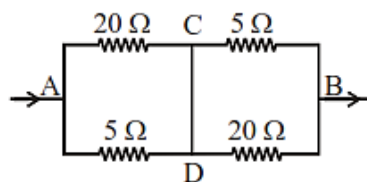
- Unit of electric power may also be expressed as
 (A) volt ampere (B) kilowatt hour
 (C) watt second (D) joule second
- Four identical resistances are joined as shown in fig. The equivalent resistance between points (A) and (B) is R_1 . The equivalent resistance between points A and C is R_2 then ratio of R_1/R_2 is:



- (A) 1 : 1 (B) 4 : 3 (C) 3 : 4 (D) 1 : 2
- An electric bulb is rated 220volt and 100watt. The resistance of the filament of the electric bulb is
 (A) 2.2 ohm (B) 2.2×10^4 ohm
 (C) 484 ohm (D) 100 ohm
 - In figure, the e.m.f. of the cell is 2V and internal resistance is negligible. The reading of the voltmeter will be

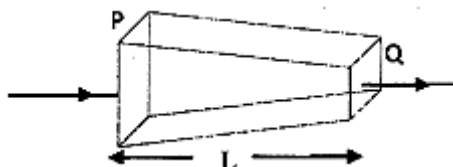


- (A) 2 volt (B) 1.33 volt
 (C) 1.60 volt (D) 0.80 volt
- When some potential difference is maintained between A and B, current I enters the network at A and leaves at B.

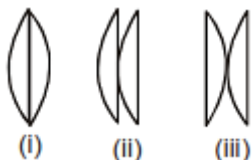


- (A) The equivalent resistance between A and B is 8Ω .
 (B) The equivalent resistance between A and B is 4Ω .
 (C) No current flows between C and D.
 (D) The equivalent resistance between A and B is 2Ω .
- A student focussed the image of a candle flame on a white screen using a convex lens. He noted down the position of the candle screen and the lens as under
 Position of candle = 12.0 cm
 Position of convex lens = 50.0 cm
 Position of the screen = 88.0 cm
 What is the focal length of the convex lens?
 (A) 19 cm (B) 20 cm
 (C) 21 cm (D) 22 cm

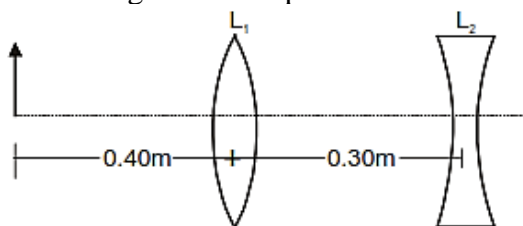
7. A conductor of length L has a varying cross section with area $2A$ at P and A at Q as shown in the adjacent figure. If it carries a steady current I , then



- (A) Current at surface P is more than surface Q .
 (B) Current at surface Q is more than surface P .
 (C) Current at surface P and surface Q are same.
 (D) None of these
8. Two mirrors are inclined at an angle 60° , an object is placed asymmetrically between them. Then number of images formed will be:
 (A) 6 (B) 5 (C) 7 (D) 9
9. Which of the following conditions are necessary for total internal reflection to take place at the boundary of two optical media?
 1. Light is passing from optically denser medium to optically rarer medium.
 2. Light is passing from optically rarer medium to optically denser medium.
 3. Angle of incidence is greater than the critical angle.
 4. Angle of incidence is less than the critical angle.
 (A) 1 and 3 only (B) 2 and 4 only (C) 3 and 4 only (D) 1 and 4 only
10. Size of image of an object by a mirror having a focal length of 20 cm is observed to be reduced to $1/3$ rd of its size. What is the nature of the image and the mirror?
 (A) Image is real and inverted. Mirror is concave.
 (B) Image is real and inverted. Mirror is convex.
 (C) Image is virtual and inverted. Mirror is concave.
 (D) Information insufficient.
11. Two identical plano-convex lenses can be combined in three ways, as shown. The ratio of the focal lengths of these combinations will be:

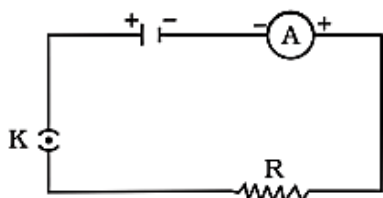


- (A) 2 : 2 : 1 (B) 1 : 1 : 1 (C) 1 : 2 : 2 (D) 2 : 1 : 1
12. The middle colour in sunlight spectrum is :
 (A) yellow (B) green (C) blue (D) orange
13. An object is placed 0.40 m from one of the two lenses L_1 and L_2 of focal lengths 0.20 m and 0.10 m respectively as depicted in the figure. The separation between the lenses is 0.30 m.

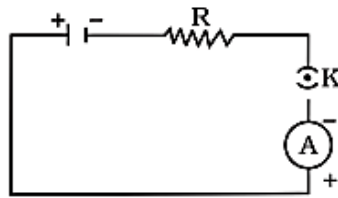


- The final image formed by this two lens system is at
 (A) 0.13 m to the right of the second lens. (B) 0.05 m to the right of the second lens.
 (C) 0.13 m to the left of the second lens (D) Infinity

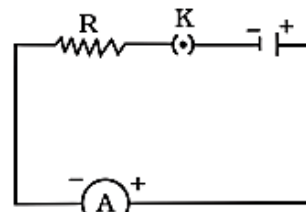
14. Which of the following can make a parallel beam of light when light from a point source is incident on it?
 (A) Concave mirror as well as convex lens
 (B) Convex mirror as well as concave lens
 (C) Two plane mirrors placed at 90° to each other
 (D) Concave mirror as well as concave lens
15. Which of the following statements is true?
 (A) A convex lens has 4 dioptre power having a focal length $+0.25$ m
 (B) A convex lens has -4 dioptre power having a focal length $+0.25$ m
 (C) A concave lens has 4 dioptre power having a focal length $+0.25$ m
 (D) A concave lens has -4 dioptre power having a focal length $+0.25$ m
16. You are given water, mustard oil, glycerine and kerosene. In which of these media a ray of light incident obliquely at same angle would bend the most?
 (A) Kerosene (B) Water (C) Mustard oil (D) Glycerine
17. A person cannot see distinctly objects kept beyond 2 m. This defect can be corrected by using a lens of power
 (A) $+0.5$ D (B) -0.5 D (C) $+0.2$ D (D) -0.2 D
18. Which of the following statements is correct regarding the propagation of light of different colours of white light in air?
 (A) Red light moves fastest
 (B) Blue light moves faster than green light
 (C) All the colours of the white light move with the same speed
 (D) Yellow light moves with the mean speed as that of the red and the violet light
19. A cell, a resistor, a key and ammeter are arranged as shown in the circuit diagrams. The current recorded in the ammeter will be



(i)



(ii)



(iii)

- (A) maximum in (i) (B) maximum in (ii)
 (C) maximum in (iii) (D) the same in all the cases
20. Which of the following represents voltage?
 (A) $\frac{\text{Work done}}{\text{Current} \times \text{Time}}$ (B) $\text{Work done} \times \text{Charge}$
 (C) $\frac{\text{Work done} \times \text{Time}}{\text{Current}}$ (D) $\text{Work done} \times \text{Charge} \times \text{Time}$

CHEMISTRY

21. Which one of the following processes involve chemical reactions?
 (A) Storing of oxygen gas under pressure in a gas cylinder
 (B) Liquefaction of air
 (C) Keeping petrol in a China dish in the open
 (D) Heating copper wire in presence of air at high temperature

22. Zinc reacts with silver nitrate to form which compounds?
 (A) $\text{Zn}(\text{NO}_3)_2 + \text{Ag}$ (B) $\text{ZnNO}_3 + \text{Ag}$
 (C) $\text{AgNO}_3 + \text{Zn}(\text{NO}_3)_2$ (D) $\text{Ag} + \text{Zn}(\text{NO}_3)_3$

23. $\text{MnO}_2 + 4\text{HCl} \longrightarrow \text{MnCl}_2 + \text{H}_2\text{O} + \text{Cl}_2$. The oxidising agent is
 (A) MnO_2 (B) HCl (C) MnCl_2 (D) $\text{Ag} + \text{Zn}(\text{NO}_3)_2$

24. $2\text{AgI}(\text{s}) \xrightarrow{\text{Sunlight}} 2\text{Ag}(\text{s}) + \text{I}_2(\text{g})$
 The colour of iodine is
 (A) Green (B) Purple (C) Brown (D) Pink

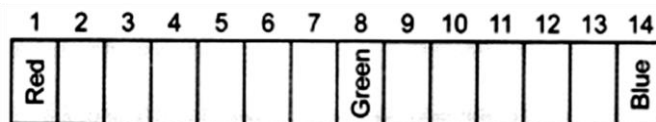
25. Which of the following is neutral salt?
 (A) NH_4Cl (B) $\text{CH}_3\text{COONH}_4$ (C) CH_3COONa (D) Na_2CO_3

26. Sodium carbonate is a basic salt because it is a salt of
 (A) strong acid and strong base. (B) weak acid and weak base.
 (C) strong acid and weak base. (D) weak acid and strong base.

27. $\text{Fe}_2\text{O}_3 + 2\text{Al} \longrightarrow \text{Al}_2\text{O}_3 + 2\text{Fe}$ This reaction is an example of –
 (A) Combination reaction (B) Double displacement reaction
 (C) Decomposition reaction (D) Displacement reaction

28. The substance that loses electrons is called as:
 (A) oxidizing agent (B) reducing agent
 (C) catalyst (D) none of above

29. Equal volumes of hydrochloric acid and sodium hydroxide solutions of same concentration are mixed and the pH of the resulting solution is checked with a pH paper. What would be the colour obtained? (You may use colour guide given in figure)

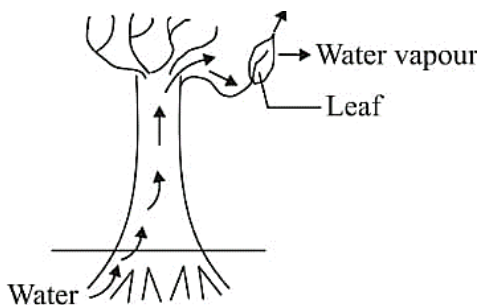


- (A) Red (B) Yellow (C) Yellowish green (D) Blue
30. In the following reaction, identify the products $\text{Na}_2\text{CO}_3(\text{aq}) + 2\text{HCl}(\text{aq}) \longrightarrow$ _____
 (A) $\text{NaCl} + \text{H}_2\text{O}$ (B) $\text{H}_2\text{O} + \text{CO}_2$
 (C) $\text{Na}_2\text{CO}_3 + \text{CO}_2 + \text{H}_2\text{O}$ (D) $\text{NaCl} + \text{CO}_2 + \text{H}_2\text{O}$
31. At what temperature is gypsum heated to form Plaster of Paris?
 (A) 35°C (B) 70°C (C) 80°C (D) 100°C
32. Which of the following acid is used by goldsmiths for cleaning gold and silver ornaments?
 (A) HF (B) H_2SO_4 (C) HNO_3 (D) H_3PO_4
33. Common salt besides being used in kitchen can also be used as the raw material for making
 (i) washing soda (ii) bleaching powder
 (iii) baking soda (iv) slaked lime
 (A) (i) and (ii) (B) (i), (ii) and (iv) (C) (i) and (iii) (D) (i), (iii) and (iv)
34. Which of the following oxide(s) of iron would be obtained on prolonged reaction of iron with steam?
 (A) FeO (B) Fe_2O_3 (C) Fe_3O_4 (D) Fe_2O_3 and Fe_2O_4

35. Which one among the following is an acidic oxide?
 (A) Na_2O (B) CO (C) CO_2 (D) Al_2O_3
36. Oxides of moderately reactive metals like Zinc, Iron, Nickel, Tin, Copper etc. are reduced by using
 (A) Aluminium as reducing agent (B) Sodium as reducing agent
 (C) Carbon as reducing agent (D) Calcium as reducing agent
37. Which of the following metals liberate hydrogen with HNO_3 ?
 (i) Cu (ii) Zn (iii) Mn (iv) Mg
 (A) (i) and (ii) (B) (ii) and (iii) (C) (iii) and (iv) (D) (i) and (iv)
38. The electronic configurations of three elements X, Y and Z are X — 2, 8; Y — 2, 8, 7 and Z — 2, 8, 2. Which of the following is correct?
 (A) X is a metal and Y is a non-metal (B) Y is a metal and Z is a non-metal
 (C) X is noble gas and Z is a non-metal (D) Y is a non-metal and Z is a metal
39. Which of the following metals exist in their native state in nature?
 (i) Cu (ii) Au (iii) Zn (iv) Ag
 (A) (i) and (ii) (B) (ii) and (iii) (C) (ii) and (iv) (D) (iii) and (iv)
40. The brown gas evolved on heating of copper nitrate is
 (A) O_2 (B) NO_2 (C) N_2 (D) NO

BIOLOGY

41. The most commonly occurring chlorophyll is
 (A) Carotene (B) Chlorophyll C (C) Xanthophylls (D) Chlorophyll a
42. Food is pushed forward in alimentary canal due to
 (A) Zig-zag movement (B) Peristaltic movement
 (C) Both (A) and (B) (D) Downward movement
43. During aerobic respiration, the complete oxidation of 1 glucose molecule produces:
 (A) 38 ATP molecules (B) 30 ATP molecules
 (C) 12 ATP molecules (D) 32 ATP molecules
44. Observe the following diagram and identify the process and its significance from the following options:



- (A) Evaporation: maintains water contents in leaf cells.
 (B) Transpiration: creates a suction force which pulls water inside the plant.
 (C) Excretion: helps in excreting out waste water from the plant.
 (D) Translocation: helps in transporting materials from one cell to another.

45. In the cardiac cycle, Diastole is
 (A) Number of heart beat per minute
 (B) Relaxation period after contraction of heart
 (C) The forceful pumping action of heart
 (D) Contraction period after relaxations heart
46. Excretion in plants mainly involves the removal of:
 (A) Solid wastes (B) Gaseous wastes (C) Liquid wastes (D) Hormonal wastes
47. The correct path taken by urine after its production is
 (A) Kidney → Ureter → Bladder → Urethra
 (B) Kidney → Bladder → Urethra → Ureter
 (C) Kidney → Urethra → Bladder → Ureter
 (D) Kidney → Ureter → Urethra → Bladder
48. Apical dominance means :-
 (A) Suppression of growth of apical bud by axillary buds
 (B) Suppression of growth of axillary buds by the presence of apical bud
 (C) Stimulation of growth of axillary buds by removal of apical bud
 (D) Inhibition of growth of axillary buds by removal apical bud
49. In plants the role of cytokinin is:
 (A) Promote cell division.
 (B) Wilting of leaves.
 (C) Promote the opening of stomatal pore
 (D) Help in the growth of stem.
50. **Statement I** : Tropic movements are the movements of a plant towards or away from the direction of a stimulus.
Statement II : Phototropism is the bending of shoot towards light.
 (A) Both statements are true (B) Both statements are false
 (C) Statement I is true. Statement II is false (D) Statement I is false. Statement II is true
51. Walking in a straight line and riding a bicycle are the activities which are possible due to a part of the brain. Choose the correct location and name of this part from the given table:

Part of the Brain	Name
(A) Fore brain	Cerebrum
(B) Mid brain	Hypothalamus
(C) Hind brain	Cerebellum
(D) Hind brain	Medulla

52. Arachnoid is
 (A) Inner meninx (B) Middle meninx (C) Outer meninx (D) Inner epithelial lining
53. In human body which is the master gland?
 (A) Adrenal (B) Pancreas (C) Pituitary (D) None
54. Select the mismatched pair:
 (A) Adrenal gland – Adrenaline
 (B) Thyroid gland – Thyroxine
 (C) Pituitary gland – Insulin
 (D) Ovary – Estrogen

55. Bryophyllum can be propagated vegetatively by-
(A) Stem (B) Root (C) Leaf (D) Flower
56. Double fertilization is
(A) Fusion of two male gametes with egg
(B) Fusion of one male gamete with egg and the other male gamete with the polar bodies
(C) Both are correct
(D) Both are incorrect
57. The genetic information in most of the organisms is stored in -
(A) DNA (B) RNA (C) Ribosome (D) ER
58. Receptive part of carpel is
(A) Placenta (B) Ovary (C) Stigma (D) Style
59. Condom is a method of birth control that falls under the following category
(A) Surgical method (B) Hormonal method
(C) Mechanical method (D) Chemical method
60. The thick lining in the uterus and rich supply of blood helps in the nourishment of
(A) Egg (B) Embryo (C) Sperm (D) Zygote

